

Risk governance

in the Chemicals Strategy for Sustainability

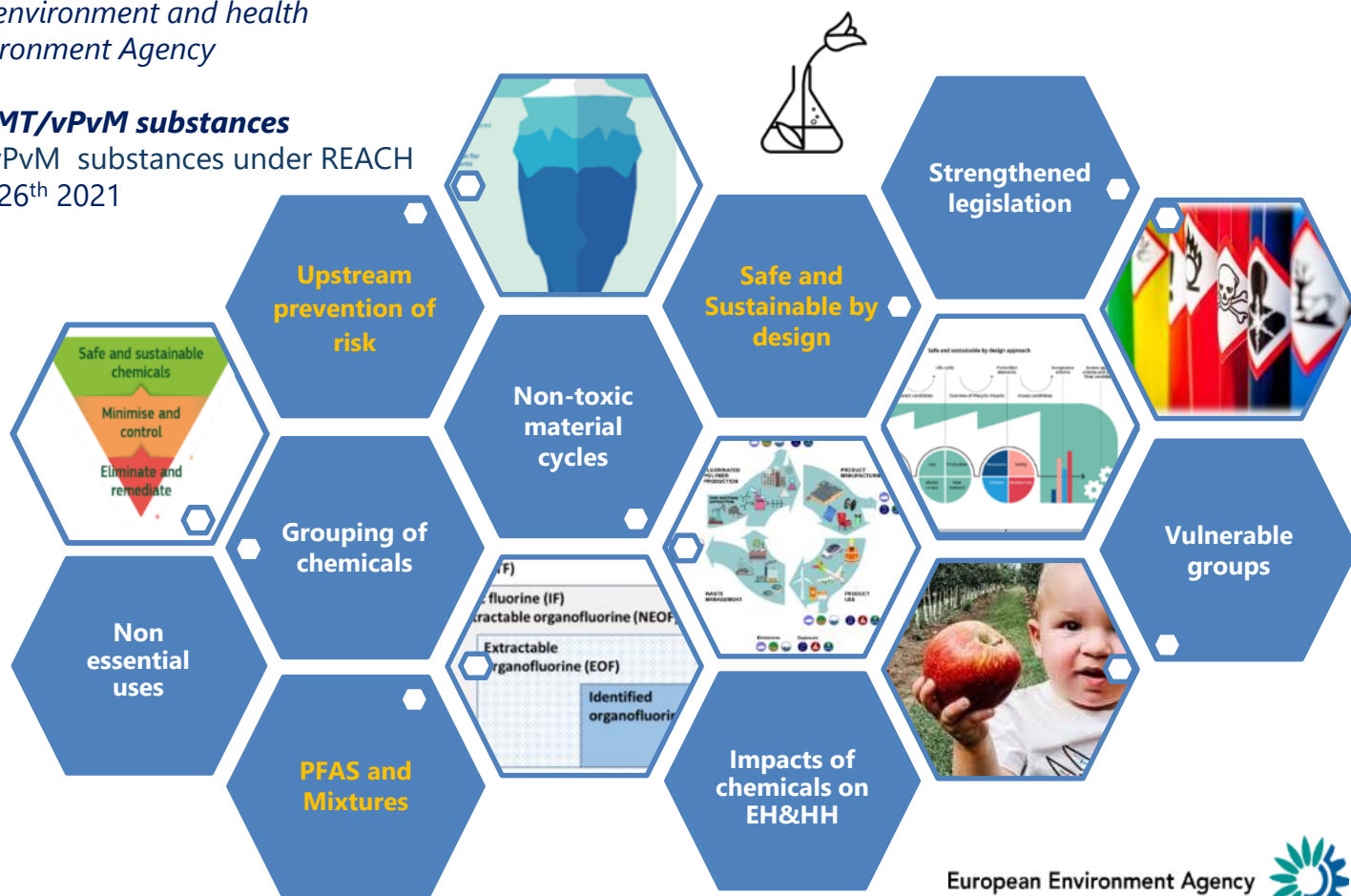
– example of PM substances and Safe and Sustainable by Design

Xenia Trier, Ph.D.

*Expert on chemicals, environment and health
European Environment Agency*

3rd workshop on PMT/vPvM substances

Getting control of PMT and vPvM substances under REACH
March 26th 2021



Concerns with PM substances

- Risk assessment cannot keep up with diversity of chemicals and uses
- Pollution occurs along lifecycles
- Persistent chemicals that do not mineralise
 - => accumulation
 - => exposure (E) increases
 - => risk (R) increases: $R = H \times E$
- Pollution spreads rapidly, across boundaries
- **Upstream prevention most effective to avoid harm to planet and people**



Read our Report [Europe's environment – State and Outlook](https://www.eea.europa.eu/publications/soer-2020/chapter-10_soer2020-chemical-pollution/)

https://www.eea.europa.eu/publications/soer-2020/chapter-10_soer2020-chemical-pollution/

Glüge et al. (2020): An overview of the uses of per- and polyfluoroalkyl substances (PFAS):

<https://pubs.rsc.org/en/content/articlelanding/2020/em/d0em00291g#!divAbstract>

Cousins et al. (2019) Why is high persistence alone a major cause of concern? Environ Sci Process Impacts 22;21(5):781-792. doi: 10.1039/c8em00515j

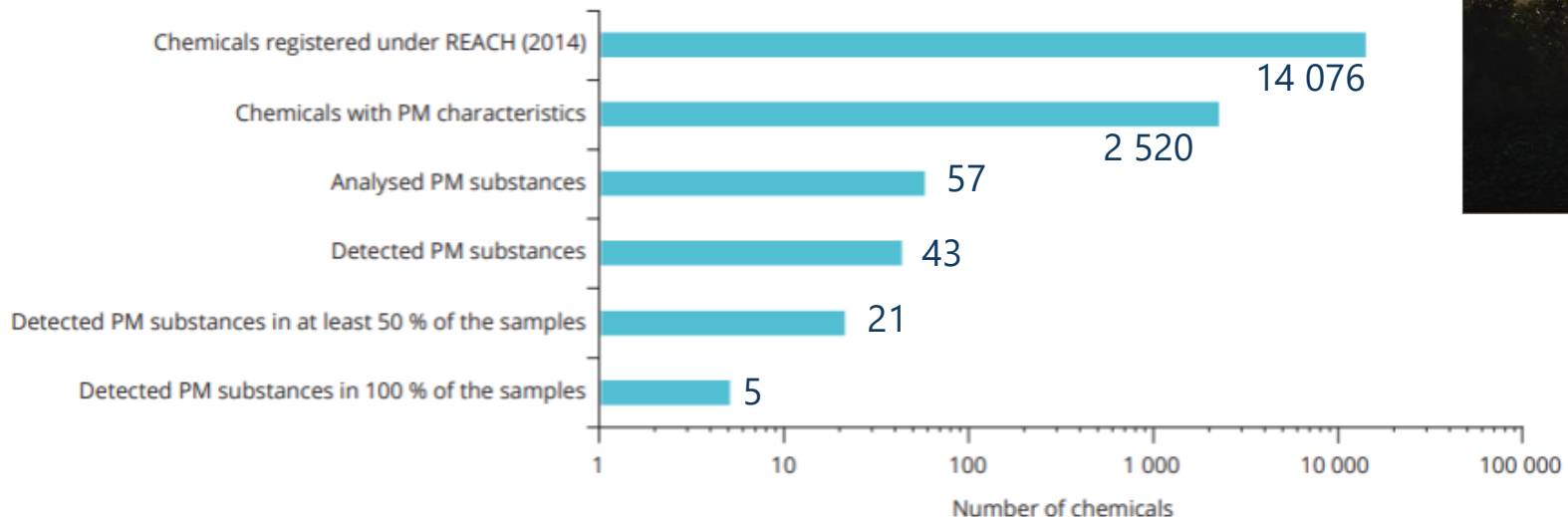
Macleod et al. (2014): Identifying chemicals that are planetary boundary threats



PM substances addressed in State and Outlook of the Environment (SOER2020)



FIGURE 10.4 Fraction of REACH chemicals that are persistent and mobile and found in water



Note: The scale is logarithmic, PM substances classified as persistent and mobile.

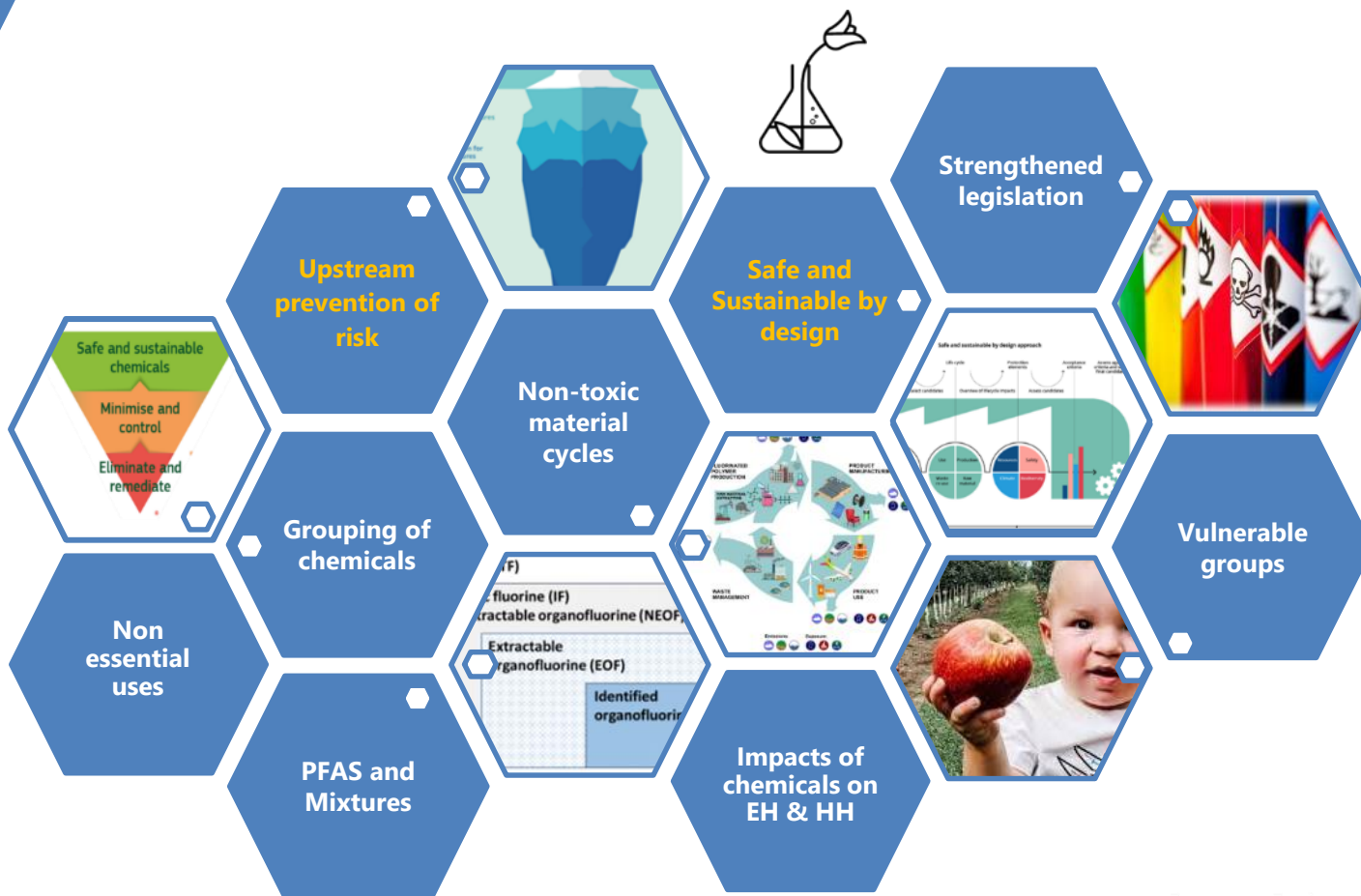
Sources: Schulze et al. (2018, 2019), Brendel et al. (2018); Arp et al. (2017); Arp and Hale (forthcoming).



Avoid harm
to planet
and people

Innovation
in SSBD

The EU Chemicals Strategy for Sustainability – focus areas



Why Safe and Sustainable by Design?



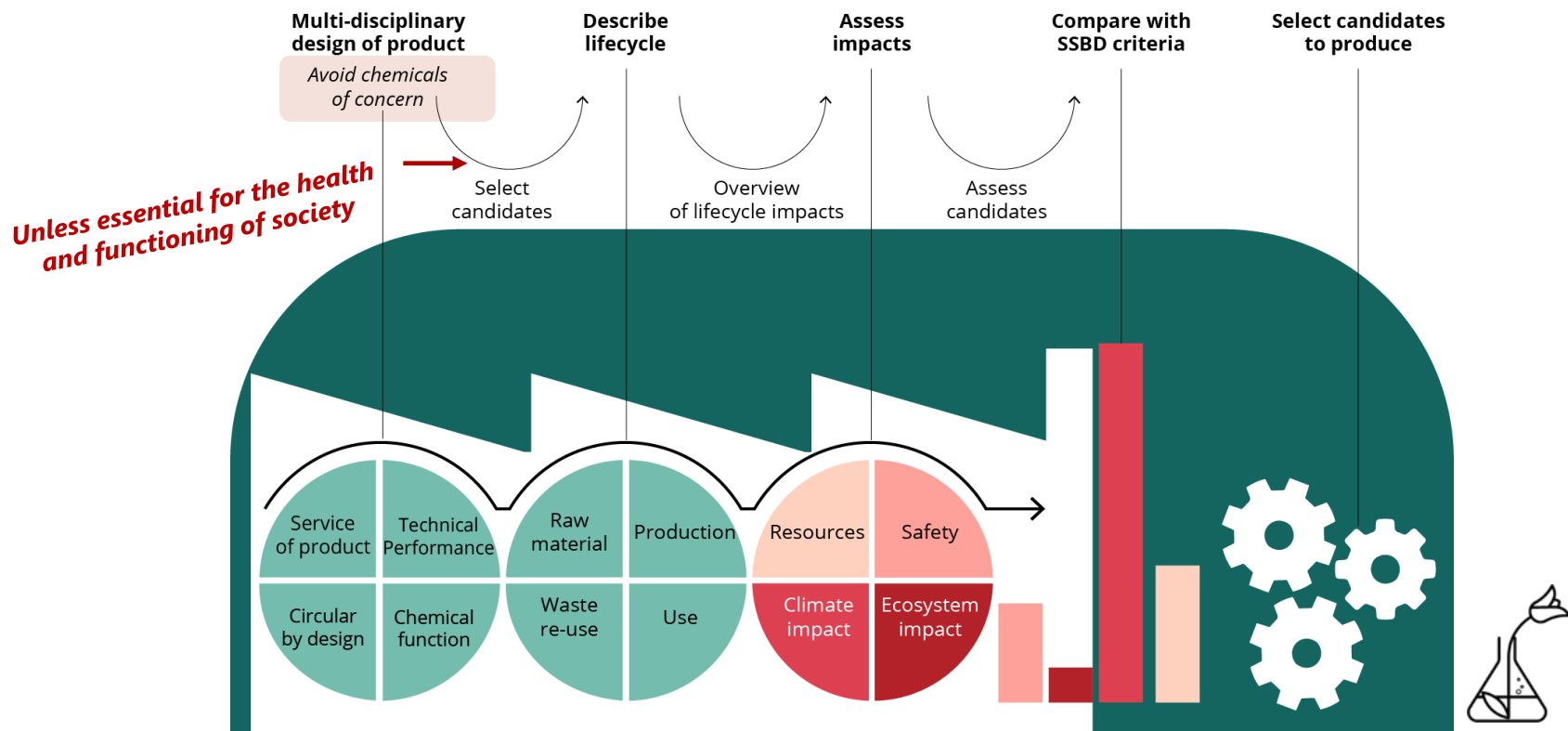
- **European Green Deal:**
Prevent pollution, environmental degradation, biodiversity loss, resource depletion & climate change - at the same time
- **Chemical Strategy for Sustainability:**
 - **Aim:** Avoid harm to planet and the people.
 - ⇒ *harm is an absolute goal, from all chemicals and stressors*
 - ⇒ *prevention of harm is most effective and manageable*
 - ⇒ *avoid the use of groups/classes of substances of concern*
 - **Tool:** Prevent pollution by Industrial transition to SSBD

Design phase

- **Premarket approach:** applied by industry in design phase
- **More flexibility** in how to deliver a *service* of a product/material, e.g. service of 'keeping dry', 'preventing fires', 'clean surfaces' etc.
- **Lifecycles** of chemicals/materials/products/processes considered



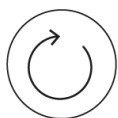
Safe and Sustainable by Design design approach



Example: prototyping a waterproof material



Design/select a few prototypes to 'keep dry'



Describe impacts of prototypes



Calculate impacts of prototypes



Compare impacts of prototypes against criteria



Select and manufacture candidate prototype(s)



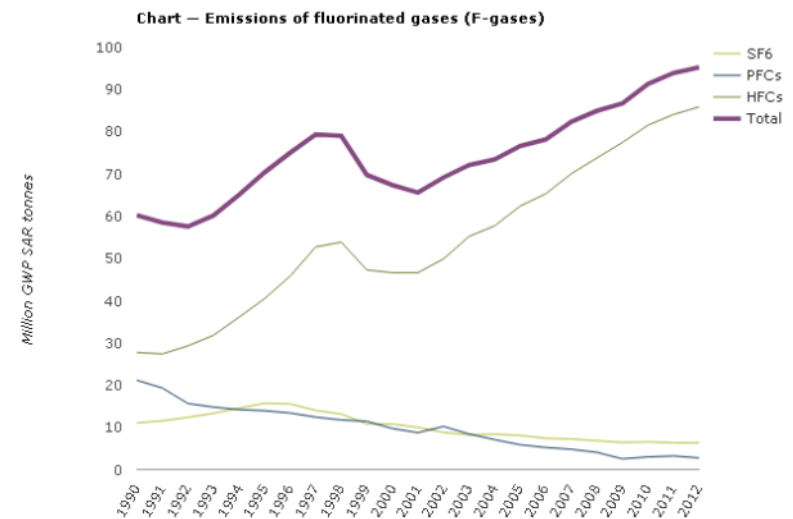
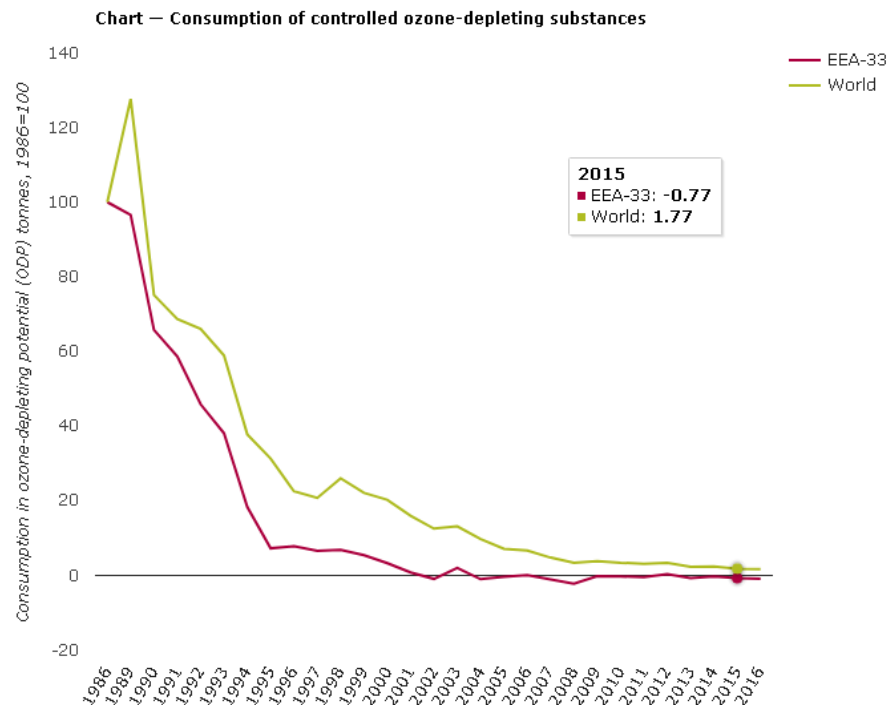
Assessing Safe and Sustainable by Design

- **Meet sustainability goals** - without compromising health of humans and environment!
- **Methodologies need input data** *on production volumes, chemical identity, uses*
- **Avoid burden shifting and create trust in SSBD** by setting minimum scores
- **Criteria for acceptable scores informed by protection goals of society**



Example: Ozone depleting substances to F-gases: Regrettable substitution

- Reduction in the use of mobile Ozone Depleting Substances (ODS) – success! 😊
- Substitution to mobile, persistent GHG F-gases => regrettable substitution - 😞
- Substitution to less GHG, mobile HFC, HFOs => Formation of persistent, mobile TFA 😞



Note: The figures show the emissions of hydrofluorocarbons (HFCs), perfluorocarbons (PFCs) and sulphur hexafluoride (SF6). Emissions are provided in metric tonnes CO₂ equivalent using the Global Warming Potential (GWP) values of the 2nd IPCC Assessment report (SAR).

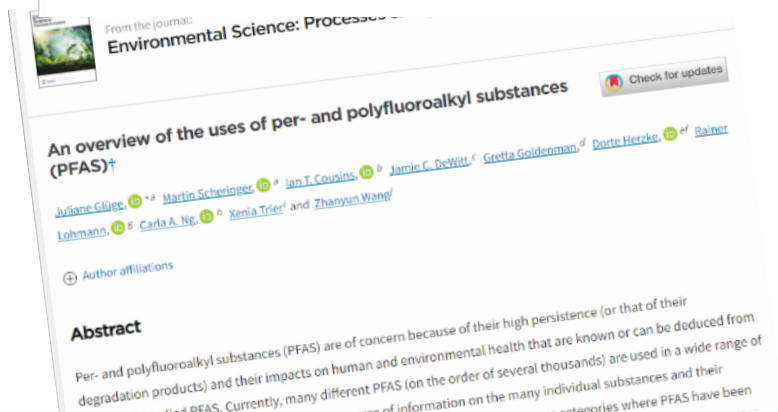
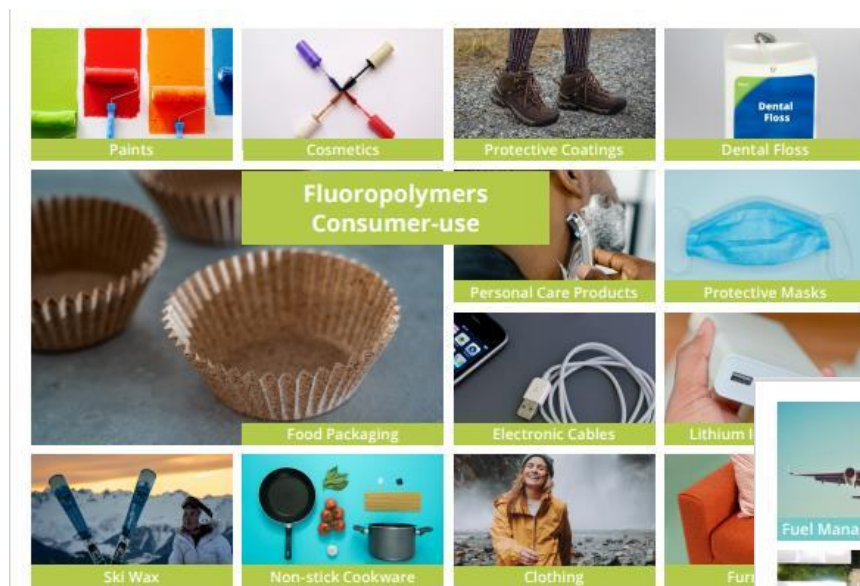
<https://www.eea.europa.eu/data-and-maps/indicators/production-and-consumption-of-ozone-2/assessment-3>

<https://www.eea.europa.eu/data-and-maps/indicators/emissions-and-consumption-of-fluorinated/assessment-2>



Example: Products for a circular economy?

Fluorinated polymers in complex materials



<https://pubs.rsc.org/en/content/articlelanding/2020/em/d0em00291g#!divAbstract>

EEA-ETC report on 'Systemic view on fluorinated polymers in a circular, low carbon and non-toxic economy' (forthcoming 2021)



Emissions of greenhouse gases (GHG) and ozone depleting substances (ODS) during production/manufacturing?

- **Synthesis of F-polymers** mostly by exchanging Cl with F atoms on hydrochlorofluorocarbon (HCFC) backbone
- **Chlorodifluoromethane** (HCFC-22) **major feedstock for tetrafluoroethylene (TFE)**
 - HCFC-22: GWP 1,760 , ODP 0.024-0.034 (0.55 in ODS Regulation)
- **Trifluoromethane** (HFC-23) major by-product in HCFC-22 production
 - GWP 12,400, ca. 2 % of HCFC-22 production volume (94,000 tons in 2018)
 - => 1,880 t HFC-23 (23 Mt CO₂-eq)
- **Many other high-GWP/-ODP substances involved in polymerization of PTFE and PVDF** and in the production of PFPEs: **emitted quantities unknown**
 - Tetrachloromethane (CTC): GWP 1,730, ODP 0.89
 - Cyclo-octafluorobutane (PFC-318): GWP 9,540
 - Tetrafluoromethane (PFC-114): GWP 6,630
 - Hexafluoroethane (PFC-116): GWP 11,100
 - 1,2-Dichloro-1,1-difluoroethane (HCFC-132b): GWP 320, ODP 0.062
 - 1-Chloro-1,1-difluoroethane (HCFC-142b): GWP 1,980, ODP 0.023-0.057



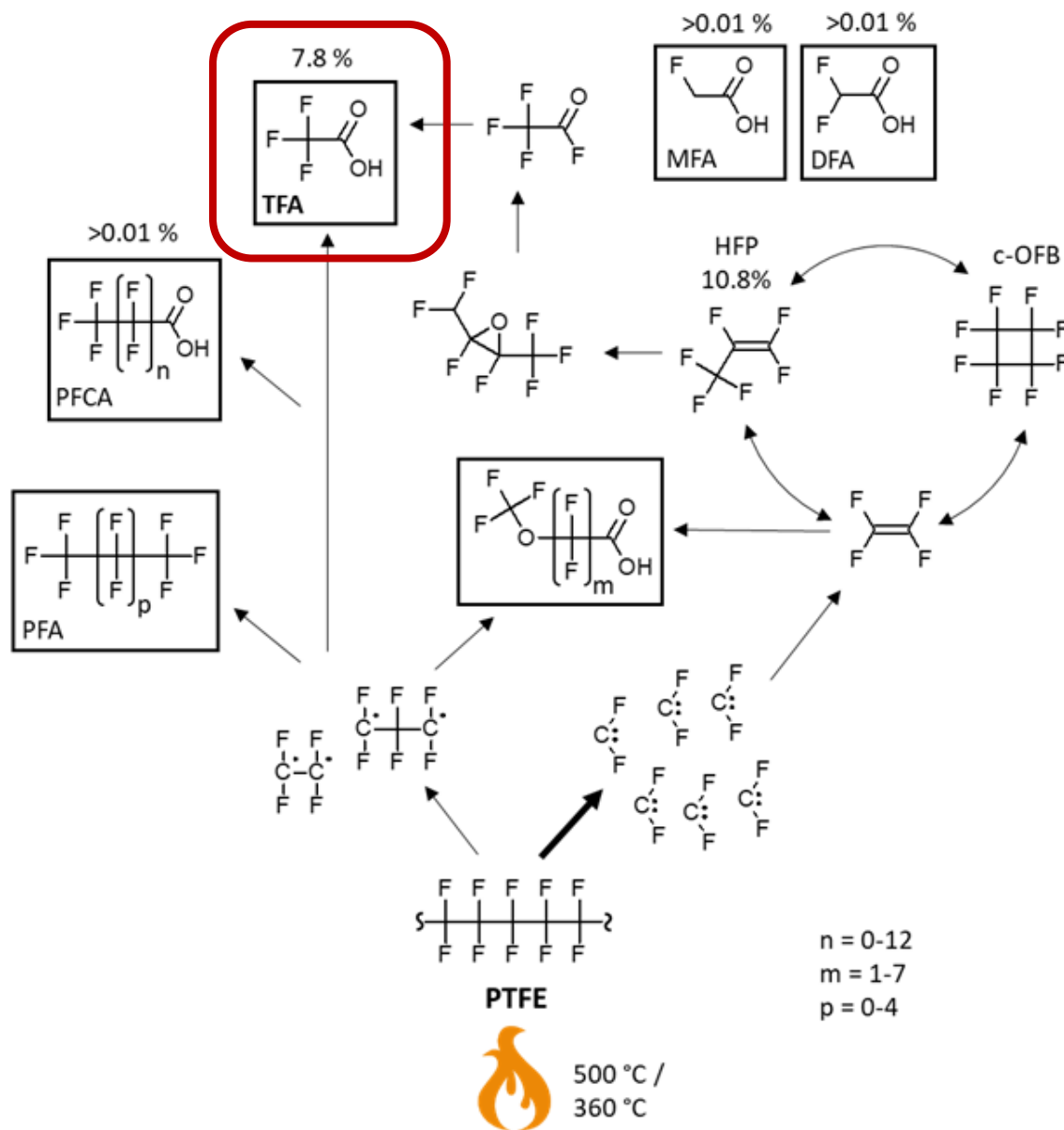
How to control safe recycling of complex products outside of Europe?



Credit: Stephen Mudge

EEA-ETC report on 'Systemic view on fluorinated polymers in a circular, low carbon and non-toxic economy' (*forthcoming 2021*)

Emissions during end of life/waste phase - incineration



- Which volatiles that are formed influenced by:

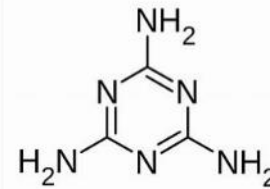
- temperature
- time heated
- O_2 availability
- energy added

- Trifluoroacetic acid (TFA) key end product: Persistent, mobile

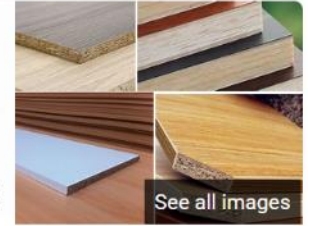


Example: Melamine

- PMT
- Used for building materials, kitchenware, etc.
- Exposure to workers during production?
- Emissions to environment: Melamine plastics depolymerize
- Service (building materials and food/liquid container) can be provided other ways
- => *is use of melamine essential for health and functioning of society?*



Melamine
Organic Compound



Example: Pharmaceuticals

– designed for environmental degradation?

- Increase in innovation and number of registrations of fluorinated pharmaceuticals
- Safe and sustainable design?

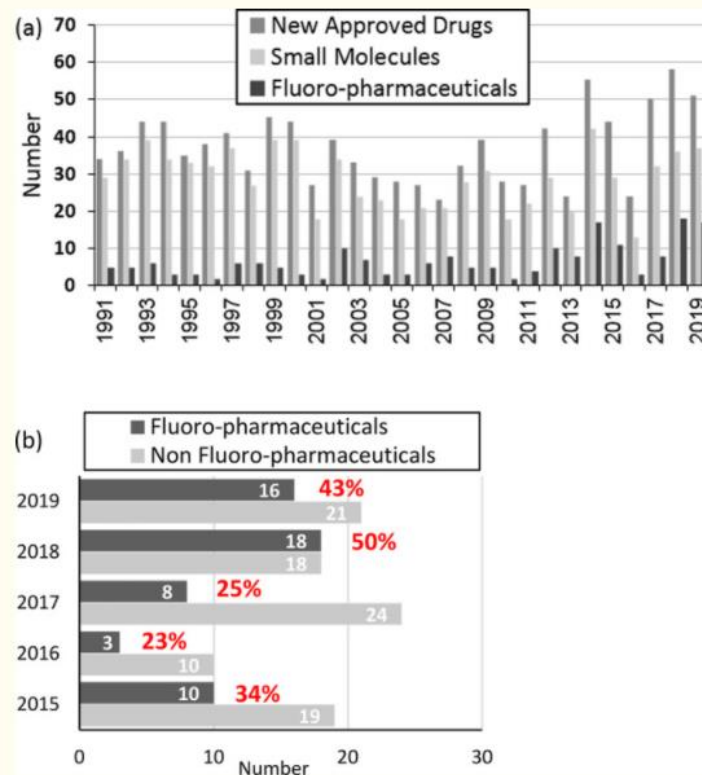
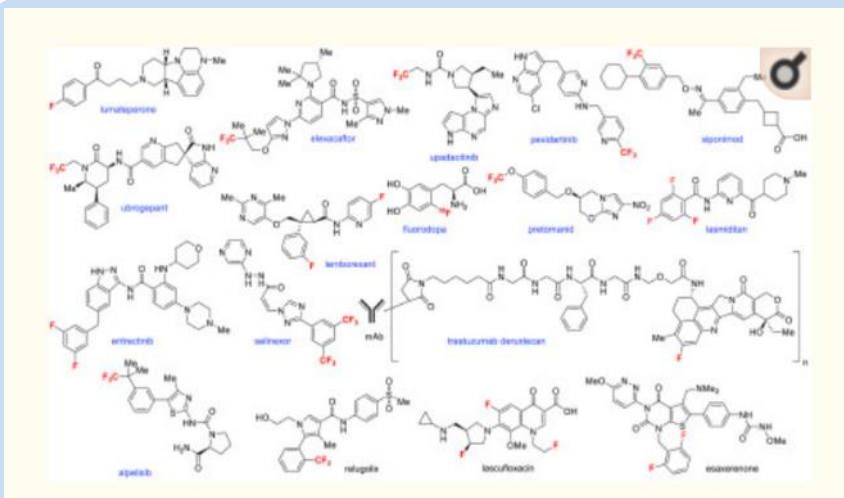


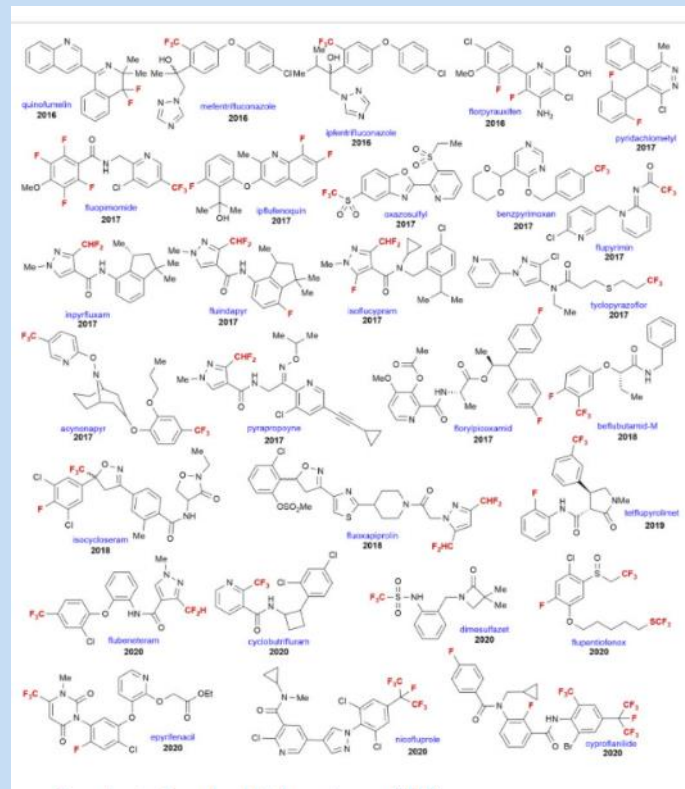
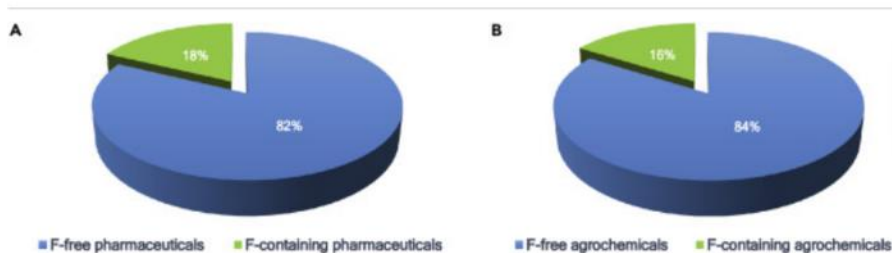
Figure 3

(a) Prevalence of fluoropharmaceuticals among globally registered drugs (1991–2019). The list of the all pharmaceuticals (1072 compounds), small-molecule drugs (839 compounds), and fluoropharmaceuticals (191 compounds) is provided (Table S2). (b) Data for small-molecule drugs over the past five years.



– designed for environmental degradation?

- **Increase in innovation and number of registrations of fluorinated pharmaceuticals**
- **Safe and sustainable design?**

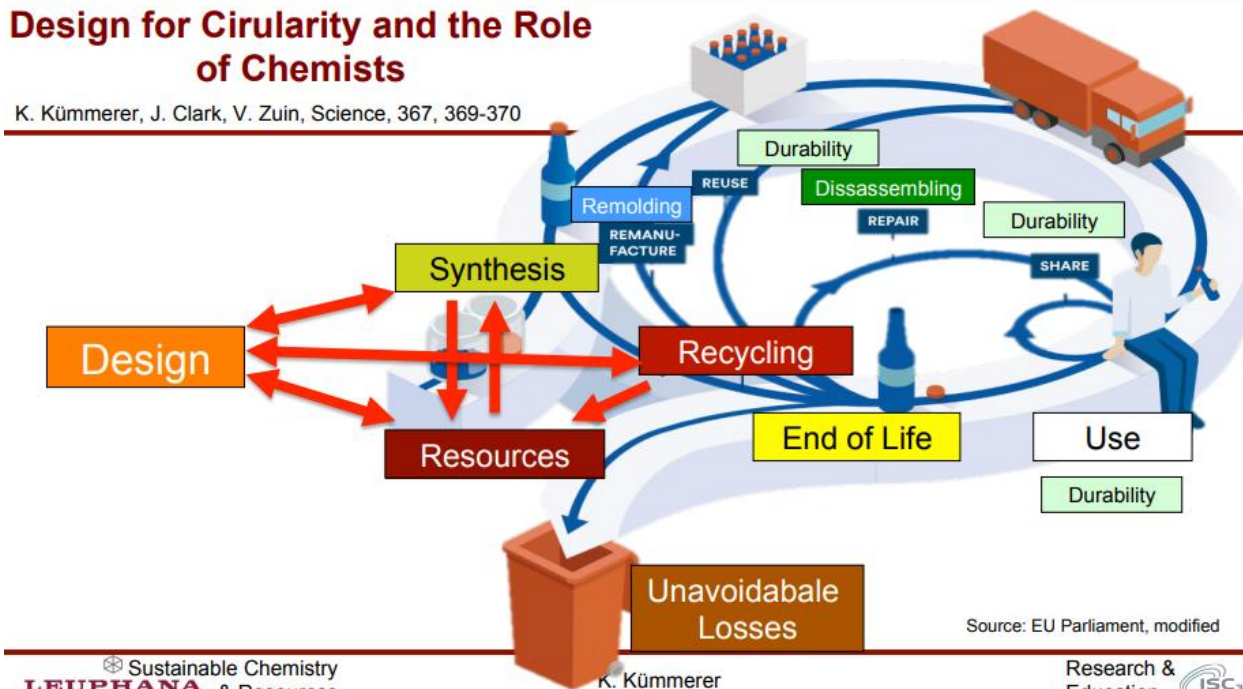


Role of chemists in designing for SSBD

- Ensure mineralization => avoid persistent chemicals
- Reduce chemodiversity and complexity of materials
- Increase reparability and seperability
- Develop methods to assess SSBD and compliance

Design for Circularity and the Role of Chemists

K. Kümmerer, J. Clark, V. Zuin, Science, 367, 369-370



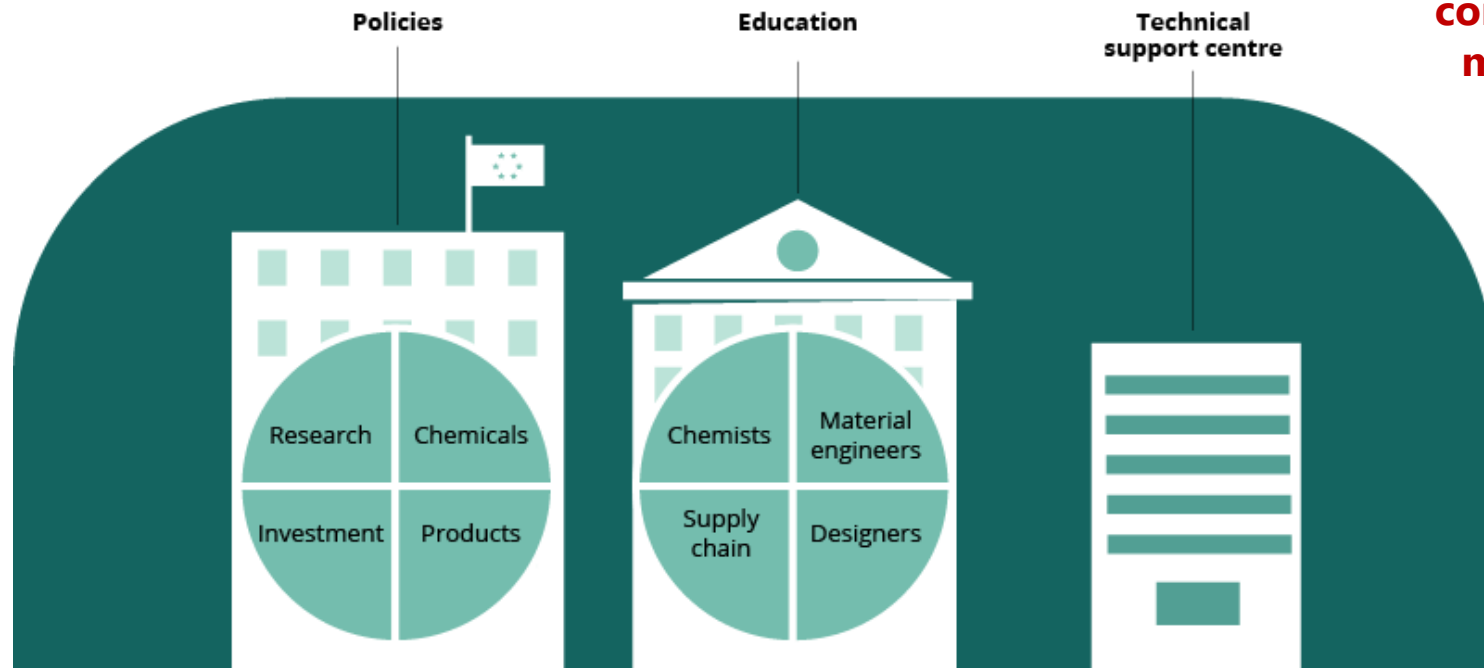
Credit: Prof. Klaus Kümmerer, Leuphana U.

European Environment Agency



Safe and Sustainable by Design – enabling environment

**Development of
compliance
methods**



**SSBD as
requirement for
Sustainable
Finance?**



Coherence
across policies



Education in SSBD
across academic/professional
disciplines



Standardised
documentation
of SSBD



Risk Governance: Assessing vs. Managing risks

- Risk assessment supports understanding of risk
- Risk management/implementation may require simpler methods for chemical classes, e.g. total PFAS

IRGC. (2017). An introduction to the IRGC Risk Governance Framework

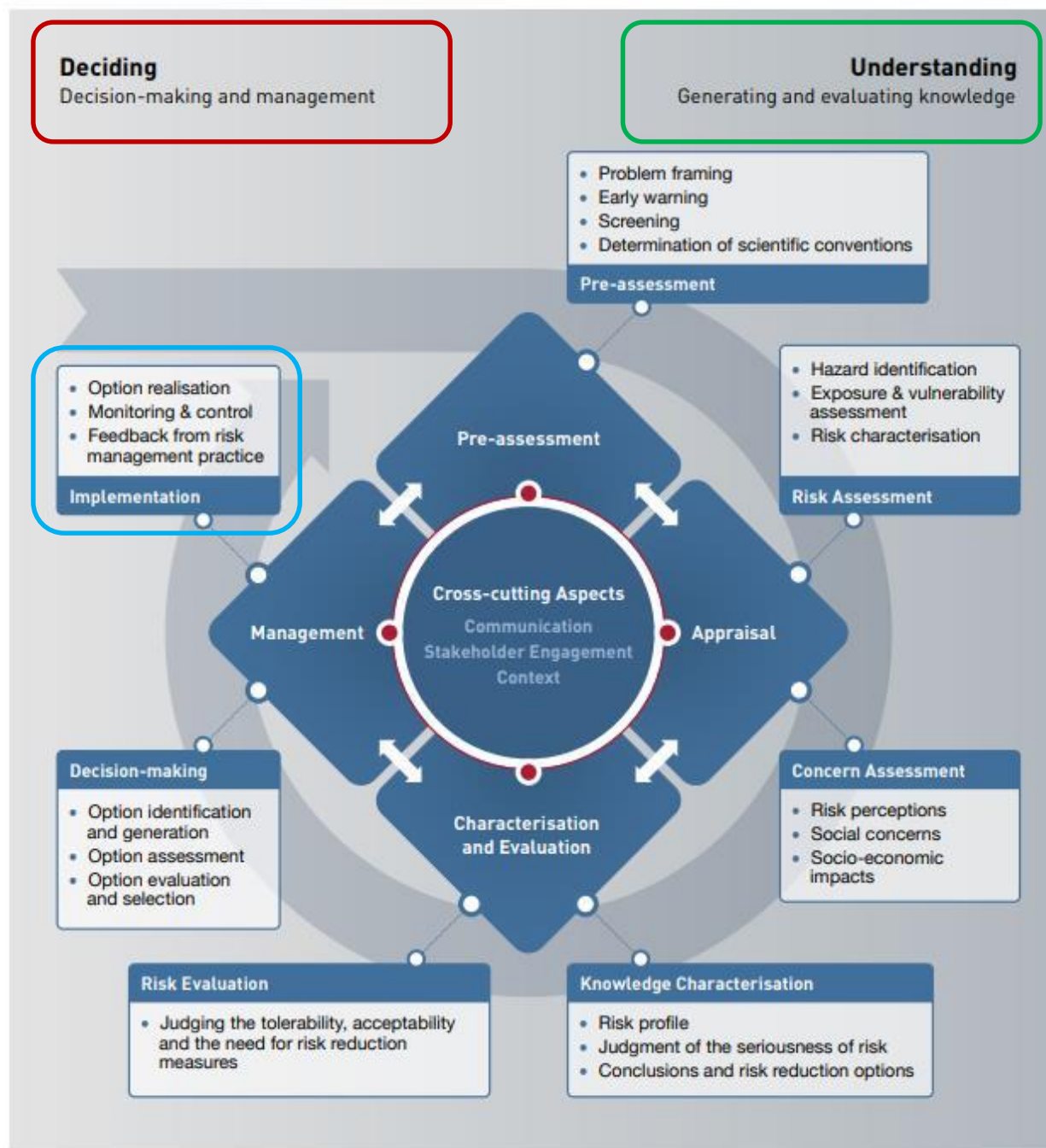


Figure 2: Detailed visual representation of the IRGC Risk Governance Framework.



Conclusions

- Cannot manage one substance at a time
- Repeated regrettable substitution, not kept in technosphere
- PMs, and their precursors to be avoided
- Complexity hampers circular economy and increase risk of pollution
- **Chemicals strategy for sustainability:**
 - Avoid harm and non-essential uses of chemicals of concern
 - Manage chemicals (e.g. PFAS) as a class
 - **Industrial transition towards innovation in safe and sustainable by design chemicals/products**, considering both safety and sustainability
- **Harmonised definitions needed for chemicals of concern**
Tools and compliance methods to test for COCs



Thank you for your attention!

xenia.trier@eea.europa.eu

Xenia Trier, Ph.D.

Chemicals, Environment and Human Health Expert
Air Pollution, Environment and Health (HSR1)
Health and Sustainable Resource Use
Kongens Nytorv 6, 1050 Copenhagen K, Denmark

eea.europa.eu

eionet.europa.eu

Phone: (+45) 33367100

Direct: (+45) 33367102



European Environment Agency

